10/764,389

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L1 STRUCTURE UPLOADED

=> d 11

L1 HAS NO ANSWERS

L1 STR

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=> s 11 full

FULL SEARCH INITIATED 14:47:30 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1406 TO ITERATE

100.0% PROCESSED 1406 ITERATIONS

SEARCH TIME: 00.00.01

L2 163 SEA SSS FUL L1

=> file caplus

COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 167.38 169.27

163 ANSWERS

FULL ESTIMATED COST

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=> s 12

L3 189 L2

=> s 13 and second dye 538711 SECOND

256811 DYE 137 SECOND DYE

(SECOND(W)DYE)

L4 0 L3 AND SECOND DYE

=> s 13 and dye

256811 DYE

L5 11 L3 AND DYE

=> dup rem 15

PROCESSING COMPLETED FOR L5

L6 11 DUP REM L5 (O DUPLICATES REMOVED)

=> d 16 bib abs hitstr 1-11

L6 ANSWER 1 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN AN 2004:885065 CAPLUS

DN 142:281518

TI Investigation on the solubilization of organic dyes and micro-polarity in AOT water-in-CO2 microemulsions with fluorinated co-surfactant by using UV-Vis spectroscopy

AU Liu, Juncheng; Ikushima, Yutaka; Shervani, Zameer

CS Supercritical Fluid Research Center, National Institute of Advanced Industrial Science and Technology, Miyagino-ku, Sendai, 983-8551, Japan

SO Journal of Supercritical Fluids (2004), 32(1-3), 97-103 CODEN: JSFLEH; ISSN: 0896-8446

PB Elsevier B.V.

DT Journal

LA English

AB It was found that the dyes thymol blue, dimidium bromide, and methyl orange, which are not soluble in pure supercrit. CO2, could be conveniently solubilized in AOT water-in-CO2 reverse microemulsions with 2,2,3,3,4,4,5,5-octafluoro-1-pentanol as co-surfactant. The solubilities of the dyes in the microemulsions were measured successfully by using a UV-visible spectroscopy method newly established in our laboratory; besides that, for a given temperature, a critical micelle pressure at which formation

AOT water-in-CO2 reverse micelles starts, was determined in term of the effect of pressure on the absorption intensity of the dyes in the microemulsions. Furthermore, the micro-polarity environment of the AOT water-in-CO2 reverse microemulsions was investigated systematically according to the shift of the solvatochromic probes methyl orange and dimidium bromide with varying water content by using UV-visible spectroscopy.

IT 518-67-2, Dimidium bromide

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(dye; solubilization of dyes and micro-polarity in AOT water-in-CO2 microemulsions with fluorinated co-surfactant)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

• Br-

RE.CNT 35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 2 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2000:908044 CAPLUS

DN 134:168628

TI Mechanisms of Solute Interfacial Transfer in Winsor-II Systems

AU Steytler, David C.; Towey, Thomas F.; Robinson, Brian H.; Atay, N. Zeynep

CS School of Chemical Sciences, University of East Anglia, Norwich Norfolk, NR4 7TJ, UK

SO Langmuir (2001), 17(2), 417-426 CODEN: LANGD5; ISSN: 0743-7463

PB American Chemical Society

DT Journal

LA English

AB The forward transfer kinetics of a water-soluble cationic dye (dimidium) across the planar interface from a conjugate aqueous phase to a water-in-oil (w/o) microemulsion phase (formed using the anionic surfactant Aerosol-OT) have been investigated by means of a rotating diffusion cell. By measurement of the solute flux as a function of rotation speed of the diffusion cell membrane, the influence of mass transport effects to and from the interface could be controlled and eliminated by extrapolation to infinite rotation speed. The rate of forward transfer was linearly proportional to the concentration of solute in

the

aqueous phase; i.e., it was not possible to saturate the aqueous side of the interface. The rate, however, was found to reach a limiting value on increasing the concentration of nano water droplets in the microemulsion phase. This is explained by a transport model in which the dye initially partitions to the aqueous side of the interface; it then enters the organic phase inside a forming water droplet. The rate of back transfer of H+ from a microemulsion droplet phase into a coexisting water phase has also been studied as a function of droplet concentration and temperature These results

extend previous measurements on the same system. It is shown that enthalpy-entropy compensation effects operate for the rate-determining step.

In

for

the proposed model for defining dynamics of interface transfer from or to an aqueous phase in Winsor-II systems, the rate-determining step is the same

forward and back transfer and is concerned with droplet coalescence with the interface.

IT 518-67-2, Dimidium bromide

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(forward transfer kinetics of dimidium bromide cationic dye and backwards transfer of H+ across planar interface from a conjugate aqueous phase to a AOT/n-heptane/water microemulsion investigated by a rotating diffusion cell)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

● Br-

RE.CNT 38 THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 3 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1999:779154 CAPLUS

DN 132:9613

TI Energy transfer hybridization assay using intercalators and lanthanide metals

IN Rabbani, Elazar; Hurley, Ian

PA Enzo Diagnostics, Inc., USA

SO U.S., 53 pp., Cont. of U.S. Ser. No. 194,215, abandoned. CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE				
ΡI	US 5998135	Α	19991207	US 1995-486053	19950607				
	US 6239271	В1	20010529	US 1999-386695	19990831				
	US 2001026921	A1	20011004	US 2001-815649	20010323				
	US 6566068	В2	20030520						
PRAI	US 1989-314995	В1	19890224						
	US 1994-194215	В1	19940209						
	US 1995-486053	A1	19950607						
	US 1999-386695	A1	19990831						

AΒ Disclosed is a nucleic acid hybridization assay composition for detecting the presence of absence of a target oligonucleotide or polynucleotide in a sample. The composition comprises: a solid matrix having at least one surface which is substituted with a first intercalator capable of binding dsDNA dsRNA, or DNA-RNA hybrids; a second intercalator, which may or may not comprise at least one fluorophore, said intercalator or said fluorophore each acting as either an energy donor or an energy acceptor; and an oligoor polynucleotide probe which is specifically hybridizable with the target oligo- or polynucleotide and has directly or indirectly bound thereto, at least one lanthanide metal chelate or at least one fluorophore, each acting as either an energy donor or an energy acceptor. The method is exemplified such that the first intercalator, the phenanthridine dye M-B 3492, is used to capture the double-stranded target-probe hybrid to the surface of the slide and a second intercalator, 9-aminoacridine, is used in solution as the energy donor. Also disclosed are a method and kit for its use.

IT 52671-19-9

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (energy transfer hybridization assay using intercalators and lanthanide metals)

RN 52671-19-9 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-(4-carboxyphenyl)-5-methyl-, chloride (9CI) (CA INDEX NAME)

• cl-

RE.CNT 32 THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 4 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1996:558603 CAPLUS

DN 125:243846

TI Luminal transport system for choline+ in relation to the other organic

cation transport systems in the rat proximal tubule. Kinetics, specificity: alkyl/arylamines, alkylamines with OH, O, SH, NH2, ROCO, RSCO and H2PO4-groups, methylaminostyryl, rhodamine, acridine, phenanthrene and cyanine compounds

- AU Ullrich, Karl J.; Rumrich, Gerhard
- CS Max Planck Inst. Biophysik, Frankfurt am Main, D-60596, Germany
- SO Pfluegers Archiv (1996), 432(3), 471-485 CODEN: PFLABK; ISSN: 0031-6768
- PB Springer
- DT Journal
- LA English

ΙT

RN

AB The efflux of [3H]choline+ from the proximal tubular lumen was measured by using the stop-flow microperfusion method. The 2-s efflux of [3H]choline+ follows kinetics with a Michaelis constant, Km = 0.18 mmol/L, maximal flux Jmax = 0.43 pmol cm-1 s-1, and a permeability term = 38.0 μm2 s-1. Replacement of Na+ by N-methyl-D-glucamine+ or Li+, or a change of luminal pH do not alter choline+ efflux. Replacement of Na+ by Cs+ inhibits 2-s choline+ (0.01 mmol/L) efflux by 22% and replacement by K+ inhibits by 49%, indicating that the elec. p.d. across the brush border membrane acts as driving force for choline+ transport. Comparing the apparent luminal inhibitory constant values for choline (apparatus Ki,l,choline+) with the chemical

structure of inhibiting substrates, it was found that the inhibitory potency of amines with high pKa values, i.e. high basicity, and of quaternary ammonium compds. (tetra-Et to tetrahexylammonium) increases with their hydrophobicity in a similar manner as was observed previously against the contraluminal N1-methylnicotinamide (NMeN+) transporter and the luminal H+/organic cation (N-methyl-4-phenylpyridinium) (MPP+) exchanger. Independently of their hydrophobicity, an increase in the inhibitory potency of the homologous series of aminoquinolines against the choline+ transporter was observed with increasing pKa values, i.e. increasing basicity, as was found previously against the 2 other organic cation transporters. A third parameter influencing the interaction with the choline+ transporter is the presence of 2 amino groups with high pKa values or one amino group and a permanent pos. charge, as is documented with the 2-ring aminostyryl and rhodamine compds., as well as 3-ring aminoacridine, amminophenanthrene and cyanine compds. Thus with the aminostyryl, pyridinium+, rhodamine, phenanthridium+ and cyanine+ dyes app.Ki,l,choline+ values of between 0.01 and 0.07 mmol/L were found. A fourth parameter influencing the choline+ transporter is the presence of an OH group on the C atom next to that bearing the N atom (as in choline+) or an ester-OCOR group (acetylcholine+, butyrylcholine+) or a thioester-SCOR-group (acetylthiocholine+, butyrylthiocholine+); or an -OP(OH)2(OR) group (glycerylphosphoryl-choline+), resulting in app. Ki, 1, choline+ values of 0.3-1.0 mmol/L. Thus, the substrates for the luminal choline+ transporter have general features in common with the luminal H+/organic cation exchanger and the contraluminal organic cation transporter, i.e. hydrophobicity and basicity. Addnl. parameters for interaction are an OH (or similar) group positioned a favorable distance from the N atom or a second amino/ammonium group in multi-ring compds. 20566-69-2, Dimidium

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); BIOL (Biological study)

(kinetics and specificity of luminal transport system for choline+ in relation to other organic cation transport systems in rat proximal tubule) 20566-69-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl- (8CI, 9CI) (CA INDEX NAME)

L6 ANSWER 5 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1995:860164 CAPLUS

DN 123:260410

TI Does the two-phase titration of surfactants require a mutagenic indicator?

AU Buschmann, N.

CS Lehrstuhl Analytische Chemie, Universitaet Muenster, Muenster, D-48149, Germany

SO Journal of the American Oil Chemists' Society (1995), 72(10), 1243 CODEN: JAOCA7; ISSN: 0003-021X

PB AOCS Press

DT Journal

LA English

AB Appropriate safety precautions are recommended when using the indicator mixture of dimidium bromide and disulphine blue, as dimidium bromide is chemical similar to ethidium bromide which is a known mutagen. When titrating anionic surfactants, dimidium bromide is found in the aqueous phase at the end of the titration; the dye should be removed by adsorption on activated charcoal which must be disposed of according to regulatory guidelines.

IT 518-67-2, Dimidium bromide
RL: ADV (Adverse effect, including toxicity); ARU (Analytical role, unclassified); ANST (Analytical study); BIOL (Biological study)
(possible mutagen; safety hazard alert in use of dimidium bromide in titration of surfactants)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

$$\begin{array}{c|c} & \text{NH2} \\ & \text{H}_2\text{N} \\ & \text{Ph} \end{array}$$

● Br-

L6 ANSWER 6 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1996:149737 CAPLUS

DN 124:235562

TI Indicator for the two-phase titration of ionic surfactants is possibly mutagenic

AU Buschmann, N.

CS CHAIR ANALYTICAL CHEMISTRY, WESTFALISCHE WILHELMS-UNIVERSITAT, Muenster, D-48149, Germany

SO Rivista Italiana delle Sostanze Grasse (1995), 72(11), 513 CODEN: RISGAD; ISSN: 0035-6808

PB Stazione Sperimentale per le Industrie degli Oli e dei Grassi

DT Journal

LA English

AΒ For the two-phase titration of anionic or cationic surfactants, the most frequently used indicator is a mixed indicator system consisting of dimidium bromide and Disulphine Blue. Recently, we noticed a similarity between ethidium bromide and dimidium bromide (I). Ethidium bromide is well known as a strong mutagenic and cancerogenic agent, reacting with DNA either by intercalation or external binding. The literature shows that I reacts with DNA in a similar way. When mixing aqueous solns. of DNA and I a change in the color of the solution can be observed, as well as a strong enhancement in the fluorescence intensity. Both the literature and the exptl. findings make it very likely that I shows mutagenic and cancerogenic properties that are similar to those of ethidium bromide. This estimation is shared by the USA National Cancer Institute. Up to now, only little is known about the ways of incorporation for both dyes. Ethidium bromide shows mutagenic and cancerogenic properties when coming in contact with the mucous membrane. The author has no information whether I will be incorporated if an aqueous solution is in contact with the intact epidermis. That would be the most probable way of incorporation when carrying out a two phase titration For that reason, the appropriate safety precautions should be taken when working with I: protection gloves and goggles and addnl., when working with the solid substance, dust mask. When titrating anionic surfactants, I will be found in the aqueous phase at the end of the titration The dye should be removed from the solution by adsorption on activated charcoal or on a polymeric resin like Amberlite XAD-16. After our experience an amount of 10 g activated charcoal is sufficient for the removal of I from one liter of the aqueous phase. IT 518-67-2, Dimidium bromide

RL: ADV (Adverse effect, including toxicity); ARG (Analytical reagent use); NUU (Other use, unclassified); REM (Removal or disposal); ANST (Analytical study); BIOL (Biological study); PROC (Process); USES (Uses) (dimidium bromide indicator for the two-phase titration of ionic surfactants is possibly mutagenic)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

$$H_2N$$
 $NH_2$ 
 $N+$ 
 $Me$ 

• Br-

L6 ANSWER 7 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1992:2847 CAPLUS

DN 116:2847

TI Quantitative determination of a DNA polymerase using intercalating dyes

IN Sutherland, John W. H.; Sheridan, Patrick James; Mezei, Louis Michael

PA Eastman Kodak Co., USA; Cetus Corp.

SO Eur. Pat. Appl., 30 pp.

CODEN: EPXXDW

DT Patent LA English FAN.CNT 1

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	PAT	ENT	NO.			KINI	)	DATE		AP	PLICAT	ION	NO.		DATE			
							-											
ΡI	EP 443823 EP 443823					A1 19910828				EP	1991-	3013	26		19910220			
						B1 19951108												
		R:	ΑT,	BE,	CH,	DE,	DK,	FR,	GB,	IT, L	I, LU,	NL,	SE					
	US	5049	490			Α		1991	0917	US	1990-	4821	37		19900	220		
	CA	2036	714			AA		1991	0821	CA	1991-	2036	714		19910	220		
	FI	9100	822			Α		1991	0821	FI	1991-	822			19910	220		
	JP	0606	2898			A2		1994	0308	JP	1991-	2167	03		19910	220		
	JΡ	3224	566			В2		2001	1029									
	AT	1300	45			E		1995	1115	ΑT	1991-	3013	26		19910	220		
	PI	PI EP EP US CA FI JP JP	PI EP 4438 EP 4438 R: US 5049 CA 2036 FI 9100 JP 0606 JP 3224	EP 443823	PI EP 443823 EP 443823 R: AT, BE, US 5049490 CA 2036714 FI 9100822 JP 06062898 JP 3224566	PI EP 443823 EP 443823 R: AT, BE, CH, US 5049490 CA 2036714 FI 9100822 JP 06062898 JP 3224566	PI EP 443823 A1 EP 443823 B1 R: AT, BE, CH, DE, US 5049490 A CA 2036714 AA FI 9100822 A JP 06062898 A2 JP 3224566 B2	PI EP 443823 A1 EP 443823 B1 R: AT, BE, CH, DE, DK, US 5049490 A CA 2036714 AA FI 9100822 A JP 06062898 A2 JP 3224566 B2	PI EP 443823 A1 1991 EP 443823 B1 1995 R: AT, BE, CH, DE, DK, FR, US 5049490 A 1991 CA 2036714 AA 1991 FI 9100822 A 1991 JP 06062898 A2 1994 JP 3224566 B2 2001	PI EP 443823 A1 19910828 EP 443823 B1 19951108 R: AT, BE, CH, DE, DK, FR, GB, US 5049490 A 19910917 CA 2036714 AA 19910821 FI 9100822 A 19910821 JP 06062898 A2 19940308 JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, L US 5049490 A 19910917 US CA 2036714 AA 19910821 CA FI 9100822 A 19910821 FI JP 06062898 A2 19940308 JP JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP 1991- EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, US 5049490 A 19910917 US 1990- CA 2036714 AA 19910821 CA 1991- FI 9100822 A 19910821 FI 1991- JP 06062898 A2 19940308 JP 1991- JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP 1991-3013 EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, NL, US 5049490 A 19910917 US 1990-4821 CA 2036714 AA 19910821 CA 1991-2036 FI 9100822 A 19910821 FI 1991-822 JP 06062898 A2 19940308 JP 1991-2167 JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP 1991-301326 EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, NL, SE US 5049490 A 19910917 US 1990-482137 CA 2036714 AA 19910821 CA 1991-2036714 FI 9100822 A 19910821 FI 1991-822 JP 06062898 A2 19940308 JP 1991-216703 JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP 1991-301326 EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, NL, SE US 5049490 A 19910917 US 1990-482137 CA 2036714 AA 19910821 CA 1991-2036714 FI 9100822 A 19910821 FI 1991-822 JP 06062898 A2 19940308 JP 1991-216703 JP 3224566 B2 20011029	PI EP 443823 Al 19910828 EP 1991-301326 19910 EP 443823 Bl 19951108 R: AT, BE, CH, DE, DK, FR, GB, IT, LI, LU, NL, SE US 5049490 A 19910917 US 1990-482137 19900 CA 2036714 AA 19910821 CA 1991-2036714 19910 FI 9100822 A 19910821 FI 1991-822 19910 JP 06062898 A2 19940308 JP 1991-216703 19910 JP 3224566 B2 20011029		

 $\ensuremath{\mathsf{AB}}$   $\ensuremath{\mathsf{A}}$  fluorometric method for determination of DNA polymerase activity is described.

19900220

Α

The method uses a single-stranded DNA template, a primer, and a dye that fluoresces when it is intercalated into double-stranded DNA but not when bound to single-stranded DNA. When Taq polymerase was assayed by this method using a bis-benzimide dye as fluorochrome a polymerase activity of 0.5 U/mL was detected in a 10 min incubation. Duplicate samples differed by  $\leq 10\%$ .

IT 20566-69-2

PRAI US 1990-482137

RL: ANST (Analytical study)

(as fluorochrome in fluorimeric assay for RNA polymerase)

RN 20566-69-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl- (8CI, 9CI) (CA INDEX NAME)

IT 518-67-2

RL: ANST (Analytical study)

(as fluorochrome in fluorimetric assay for DNA polymerase)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

ANSWER 8 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN L6

1991:8568 CAPLUS ΑN

DN 114:8568

TΙ Colorimetric determination of anionic surfactants

Orthgiess, Erhard; Dobias, Bohuslav ΑU

CS

Inst. Phys. Makromol. Chem., Univ. Regensburg, Regensburg, Germany Tenside, Surfactants, Detergents (1990), 27(4), 226-8 CODEN: TSDEES; ISSN: 0932-3414 SO

DТ Journal

LA German

AB A method for the determination of anionic surfactants such as alkyl- and alkylarylsulfates, -sulfonates, and -sulfosuccinates in aqueous solns. is based on the absorbance measurement of surfactant-dye (i.e., 3,8-diamino-5-methyl-6-phenylphenanthridinium bromide) complexes at 525 nm. The technique is simpler and faster than other colorimetric methods and can be used for the anal. of 1 + 10-4 to 5 + 10-7 M solns.

518-67-2, 3,8-Diamino-5-methyl-6-phenylphenanthridinium bromide RL: USES (Uses) ΙT

(complexing agents, in colorimetric determination of anionic surfactants)

RN 518-67-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

● Br⁻

L6 ANSWER 9 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

1988:201319 CAPLUS ΑN

108:201319 DN

TΙ Polynucleotide detection by hybridization probe

PA Enzo Biochem, Inc., USA

SO Jpn. Kokai Tokkyo Koho, 31 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

		_														
	PAT	rent	NO.			KINI	)	DATE			APE	PLICATION NO.	DATE			
			<del>-</del> -				-						 			
ΡI	JP	6216	3699			A2		1987	0720		JP	1986-295047	19861212			
	JΡ	3293	820			B2		2002	0617							
	JΡ	2002	35649	96		A2		2002	1213		JΡ	2002-19249	19861212			
	EΡ	2314	95			A2		1987	0812		EΡ	1986-117432	19861215			
	EΡ	2314	95			А3		1990	0718							
	EΡ	2314	95			В1		1999	0616							
		R:	ΑT,	CH,	DE,	FR,	GB,	IT,	LI,	SE						
	CA	1339	096			A1		1997	0729		CA	1986-525363	19861215			
	EΡ	9167	37			A2		1999	0519		ΕP	1998-118357	19861215			
	EΡ	9167	37			A3		2002	0306							
		R:	ΑT,	CH,	DE,	FR,	GB,	IT,	LI,	SE						

	AT 181366	E	19990715	AT 1986-117432	19861215
	JP 07322900	A2	19951212	JP 1995-140050	19950515
	JP 2735813	B2	19980402		
PRAI	US 1985-808757	Α	19851213		
	JP 1986-295047	A3	19861212		
	EP 1986-117432	A3	19861215		

AB Test double-stranded polynucleotides in a sample are contacted with a single-stranded polynucleotide probe consisting of polynucleotide and at least a 1st element (e.g. dye) linked to the polynucleotide via an arm (e.g. allylamine) and a 2nd element (e.g. dye) linked to the polynucleotide via a 2nd arm (the 1st and 2nd segments of the polynucleotide probe are separated by .apprx.10 other nucleotides), and characteristic changes in hybridization are measured for homogeneous detection of the test polynucleotide. The characteristic changes are changes in fluorescence or thermodn. stability. The dyes are phenanthridine, acridine, and anthracycline. The test involves transformation of the double-stranded form to a single-stranded form prior to hybridization.

IT 20566-69-2, Dimidium

RL: ANST (Analytical study)

(multilabeled polynucleotide probe containing, for polynucleotide detection, fluorescence or thermodn. changes in relation to)

RN 20566-69-2 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl- (8CI, 9CI) (CA INDEX NAME)

L6 ANSWER 10 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1986:16644 CAPLUS

DN 104:16644

TI Self-catalyzed cyclization of the intervening sequence RNA of Tetrahymena: inhibition by methidiumpropyl·EDTA and localization of the major dye binding sites

AU Tanner, N. Kyle; Cech, Thomas R.

CS Dep. Chem., Univ. Colorado, Boulder, CO, 80309, USA

SO Nucleic Acids Research (1985), 13(21), 7759-79 CODEN: NARHAD; ISSN: 0305-1048

DT Journal

LA English

The intervening sequence (IVS) excised from the rRNA precursor of T. thermophila is converted to a covalently closed circular RNA in the absence of proteins in vitro. This self-catalyzed cyclization reaction is inhibited by the intercalating dye methidiumpropyl·EDTA (MPE) (Hertzberg, R. P.; Dervin, P. B., 1982). The MPE-binding sites were localized by mapping the sites of MPE·Fe(II) cleavage of the IVS RNA. There are 3 major binding sites within the 414 nucleotide IVS RNA. Two of these sites coincide with the A·B and 9L·2 pairings. These are structural elements that are conserved in all group I introns and are implicated as being functionally important for splicing. It is proposed that interaction of the MPE with these sites is responsible for dye inhibition of cyclization. The reactions of MPE·Fe(II) with an RNA of known structure, tRNAPhe, and with the IVS RNA were studied as a function of temperature, ionic strength, and ethidium concentration Based on the

comparison of the reaction with these 2 RNAs, it is concluded that the dye is a very useful probe for structural regions of large RNAs, whereas it provides more limited structural information about the small, compact tRNA mol.

IT 83693-09-8

RL: BIOL (Biological study)

(RNA site-specific cleavage by, as structural probe)

RN 83693-09-8 CAPLUS

CN Iron, [3,8-diamino-6-[4-[13-(carboxy-κ0)-9,12-bis[(carboxy-κ0)methyl]-1-oxo-7-(oxo-κ0)-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methylphenanthridiniumato(2-)]- (9CI) (CA INDEX NAME)

IT 80082-09-3

RL: BIOL (Biological study)

(cyclization of rRNA precursor intervening sequence of Tetrahymena thermophila inhibition by, di-binding sites in relation to)

RN 80082-09-3 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[13-carboxy-9,12-bis(carboxymethyl)-1,7-dioxo-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methyl- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

L6 ANSWER 11 OF 11 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1983:434663 CAPLUS

DN 99:34663

TI Inhibition of cation-induced DNA condensation by intercalating dyes

AU Widom, Jonathan; Baldwin, Robert L.

CS Med. Cent., Stanford Univ., Stanford, CA, 94305, USA

Biopolymers (1983), 22(6), 1621-32 CODEN: BIPMAA; ISSN: 0006-3525

CODEN: BIPMAA; ISSN:

DT Journal LA English

SO

AB Several intercalating dyes inhibit the cation-induced condensation of  $\lambda$ -DNA when Co3+(NH3)6 is the condensing agent. The dyes studied are ethidium, propidium, proflavine, quinacrine, and actinomycin D. Dye-induced decondensation of intramol. condensed DNA was studied by using conditions in which Co3+(NH3)6 produces intramol. condensation without significant aggregation. Some aggregation is caused, however, during dye-induced decondensation. Dye titration curves of DNA decondensation were measured by excess light scattering to monitor decondensation and by fluorescence to monitor intercalation. All of the dyes studied act as competing cations in displacing the condensing cation Co3+(NH3)6 from the DNA. Competition occurs both in and below the transition zone for condensation. The effectiveness of a dye as a competing cation increases with its net pos. charge. Before decondensation begins, no intercalated dye can be detected, suggesting that intercalation might be incompatible with the proper helix packing needed for cation-induced DNA condensation. To test this last point, methidium-spermine was synthesized: it contains an intercalating methidium head group combined with a polyamine tail. Methidium-spermine caused  $\lambda$ -DNA condensation, but aggregation accompanies condensation, as has been found earlier for spermine and spermidine. Fluorescence and absorption spectra indicate that the methidium group is intercalated when the DNA is condensed, indicating that intercalating need not be incompatible with DNA condensation. The presence of aggregates among the condensed DNA mols. makes this last conclusion tentative. ΙT 86388-76-3

RL: BIOL (Biological study)

(cation-induced DNA condensation response to)

RN 86388-76-3 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[[[3-[[4-[(3-aminopropyl)amino]butyl]amino]propyl]amino]carbonyl]phenyl]-5-methyl-(9CI) (CA INDEX NAME)

$$_{\rm H_2N}$$
  $_{\rm Me}$   $_{\rm C-NH-(CH_2)_3-NH-(CH_2)_4-NH-(CH_2)_3-NH_2}$ 

=> => d his

(FILE 'HOME' ENTERED AT 14:41:16 ON 15 AUG 2006)

FILE 'REGISTRY' ENTERED AT 14:46:42 ON 15 AUG 2006

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L1
                     STRUCTURE UPLOADED
L2
               163 S L1 FULL
      FILE 'CAPLUS' ENTERED AT 14:47:36 ON 15 AUG 2006
L3
                189 S L2
L4
                  0 S L3 AND SECOND DYE
L5
                 11 S L3 AND DYE
L6
                 11 DUP REM L5 (O DUPLICATES REMOVED)
=> s 13 and label?
          436402 LABEL?
L7
               13 L3 AND LABEL?
=> s 17 not 16
rac{1}{8}
               11 S L6
L9
               13 L7 NOT L8
=> d 19 bib abs hitstr 1-13
L9
      ANSWER 1 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN
      2005:1329208 CAPLUS
ΑN
      144:65144
DN
      Detection of DNA mismatches and oxidative lesions by converting internal
TI
      3'-phosphate termini to 3'-hydroxy termini and labeling
ΙN
      Barton, Jacqueline K.; Hart, Jonathan
PΑ
      California Institute of Technology, USA
SO
      PCT Int. Appl., 40 pp.
      CODEN: PIXXD2
DT
      Patent
LA
      English
FAN.CNT 1
      PATENT NO.
                                 KIND
                                          DATE
                                                           APPLICATION NO.
                                                                                          DATE
                                 ____
                                          _____
                                                           -----
                                                                                          _____
PΙ
      WO 2005121375
                                 A2
                                          20051222
                                                          WO 2005-US20101
                                                                                          20050607
           W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
                CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA,
                 ZM, ZW
           RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM,
                 AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML,
                 MR, NE, SN, TD, TG
      US 2006014181
                                  Α1
                                          20060119
                                                           US 2005-147805
                                                                                          20050607
PRAI US 2004-577900P
                                  Ρ
                                          20040607
      MARPAT 144:65144
OS
GΙ
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Τ

Described herein are methods for directly labeling the 3'-phosphate end associated with photocleavage at a mismatch site. Because internal 3'-phosphate termini on DNA duplexes are also associated generally with oxidative lesions, these methods provide a general strategy also for labeling and therefore detecting the frequency of oxidative DNA lesions. Labeling using terminal transferase or nontemplated DNA polymerization is also envisaged, where using either of these activities it is possible to tag a damaged site, after removal of the 3'-phosphate, with polynucleotide tails. Such polynucleotide tails in turn can be used as primer binding sites for use in PCR. In one embodiment, a method of detecting internal 3'-phosphate termini in a nucleic acid duplex from at least one is envisaged including contacting the nucleic acid duplex with an agent to convert an internal 3'-phosphate termini to 3'-hydroxy termini, extending 3'-hydroxy termini present in the duplex by nontemplate dependent DNA polymerization, amplifying the extended product of the resulting products and identifying a nucleotide sequence-dependent feature in the resulting amplified products, where the identified feature in amplified products correlates with the presence of internal 3'-phosphate termini. Further, the converting step may include, but is not limited to, contacting the internal 3'-phosphate termini with T4-polynucleotide kinase (4-TNK) and the nucleic acid duplex containing a mismatched or damaged base. Moreover, the method includes, but is not limited to, contacting the duplex with an AP lyase (e.g., APN1), and in a related aspect, the nontemplate polymerization is carried out with TAQ polymerase, terminal deoxynucleotide transferase (TdT), or DNA polymerase Mu (Pol  $\mu$ ). In another related aspect, annealed nucleic acids may be obtained from more than one sample and nicks may be generated in the annealed product with an agent that cleaves mismatched or damaged nucleotides to generate internal 3'-phosphate termini. Moreover, at least one of the sample nucleic acid duplexes may include an annealed nucleic acid probe. In one aspect, the agent is a hindered intercalating compound of the formula Rh(R1)(R2)(R3)3+, where R1 and R2 are each independently aryl, heteroaryl, substituted aryl or substituted heteroaryl of 1 to 5 rings, and R3 is a group of the formula (I) wherein x and z are each independently an integer from 1 to 4 and y is an integer from 1 to 2, and R4, R5, and R6 are each independently H-, halo, HO-, H2N-, CN-, O2N-, HS-, O3S-, O3SO-, -COOH, -CONH2, RRO-, RNH-, RaRbN-, RO3S-, RO3SO-, -COOR, -CONHR, or -CONRaRb, where R, Ra, and Rb are each independently lower alkyl, cycloalkyl, lower alkenyl, lower alkynyl, or phenol, or two R4, R5, or R6 together form a fused aryl ring, wherein the compound intercalates between bases in the presence of polynucleotide damage or error and does not intercalate between bases in the absence of damage or error. In a related aspect, the agent is  $\Delta$ - or  $\Lambda$ -Rh(bpy)2(chrysi)3+, where cleaving comprises photocleavage. In a related aspect, the mismatch is allelic and may include, but is not limited to, a single nucleotide polymorphism (SNP). In another aspect, the damage is a DNA lesion from oxidative stressor exposure, UV light exposure, or adduct formation. In one embodiment, a method of identifying mismatches in a sample nucleic acid duplex is envisaged, including producing nicks in the duplex with an agent that cleaves mismatched nucleotides to generate internal 3'-phosphate termini, extending the internal 3'-phosphate termini by nontemplate dependent DNA polymerization, amplifying the extended product, and determining a nucleotide sequence-dependent feature of the resulting amplified products, where differentiation of the feature between amplified products correlates with the presence of a mismatched base. Sensitive methods were developed to quantitate the frequency of mismatches. By labeling the site of mismatch photocleavage, either by fluorescence, radioactivity, or polymerization,

quantitation of mismatch cleavage and hence the frequency of mismatches can be achieved. Herein described are methods for directly labeling the 3'-phosphate end associated with photocleavage at a mismatch site.

ΙT 83693-09-8

AΒ

RL: RGT (Reagent); RACT (Reactant or reagent)

(to convert an internal 3'-phosphate termini to 3'-hydroxy termini,; detection of DNA mismatches and oxidative lesions by converting internal 3'-phosphate termini to 3'-hydroxy termini and labeling)

RN 83693-09-8 CAPLUS

Iron,  $[3,8-diamino-6-[4-[13-(carboxy-\kappa0)-9,12-bis[(carboxy-$ CN  $\kappa$ O)methyl]-1-oxo-7-(oxo- $\kappa$ O)-2,6,9,12-tetraazatridec-1yl]phenyl]-5-methylphenanthridiniumato(2-)]- (9CI) (CA INDEX NAME)

$$\begin{array}{c} 0 \\ 0 \\ 0 \\ \end{array}$$

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L9
     ANSWER 2 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN
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ΑN 2005:902740 CAPLUS

143:263095 DN

TISelective high-affinity polydentate ligands and methods of making such

ΙN Denardo, Sally; Denardo, Gerald; Rodney, Balhorn

PΑ The Regents of the University of California, USA

PCT Int. Appl., 106 pp. SO CODEN: PIXXD2

DT Patent

LA English

באו כאיד 1

FΑ	$N \cdot C$	PATENT NO.																			
		PAT	ENT I	NO.			KIN	)	DATE		1	APPL	ICAT:	ION I	NO.		DATE				
ΡI		WO 2005077065 WO 2005077065					A2 200508 A3 200512				Ţ	WO 2	005-	US41	34		20050208				
			W: RW:	CN, GE, LK, NO, TJ, BW, AZ, EE,	CO, GH, LR, NZ, TM, GH, BY, ES,	AL, CR, GM, LS, OM, TN, GM, KG,	AM, CU, HR, LT, PG, TR, KE, KZ, FR,	AT, CZ, HU, LU, PH, TT, LS, MD, GB,	AU, DE, ID, LV, PL, TZ, MW, RU, GR, BF,	AZ, DK, IL, MA, PT, UA, MZ, TJ,	DM, IN, MD, RO, UG, NA, TM, IE,	DZ, IS, MG, RU, US, SD, AT, IS,	EC, JP, MK, SC, UZ, SL, BE, IT,	EE, KE, MN, SD, VC, SZ, BG, LT,	EG, KG, MW, SE, VN, TZ, CH, LU,	ES, KP, MX, SG, YU, UG, CY, MC,	FI, KR, MZ, SK, ZA, ZM, CZ, NL,	GB, KZ, NA, SL, ZM, ZW, DE, PL,	GD, LC, NI, SY, ZW, AM, DK, PT,	SM	
					•	•	TD,														
חח		US 2006084115						2006 2004		1	US 2005-55181					20050209					
PK	(AT	US 2004-543444P					Ρ.		ZUU4	ロスロタ											

20040209

AΒ This invention provides novel polydentate selective high affinity ligands (SHALs) that can be used in a variety of applications in a manner analogous to the use of antibodies. SHALs typically comprise a multiplicity of ligands that each bind different region of the target mol. The ligands are joined directly or through a linker thereby forming a polydentate moiety that typically binds the target mol. with high selectivity and avidity. 80082-09-3, Methidiumpropyl EDTA

ITRL: RCT (Reactant); RACT (Reactant or reagent)

(selective high-affinity polydentate ligands and methods of making such)

RN80082-09-3 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[13-carboxy-9,12-bis(carboxymethyl)-1,7dioxo-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methyl- (9CI) (CA INDEX NAME)

PAGE 1-A

$$_{\text{H}_{2}\text{N}}^{\text{NH}_{2}}$$
 $_{\text{Me}}^{\text{CH}_{2}-\text{CO}_{2}\text{H}}$ 
 $_{\text{C}}^{\text{CH}_{2}-\text{CO}_{2}\text{H}}$ 
 $_{\text{C}}^{\text{CH}_{2}-\text{CO}_{2}\text{H}}$ 
 $_{\text{C}}^{\text{CH}_{2}-\text{CH}_$ 

PAGE 1-B

ANSWER 3 OF 13 , CAPLUS COPYRIGHT 2006 ACS on STN 2003:771377 CAPLUS L9

ΑN

DN 139:288630

TΙ Labeling reagents and labeled targets, target labeling processes and other processes for using same in nucleic acid determinations and analyses

ΙN Stavrianopoulos, Jannis G.; Rabbani, Elazar

PΑ Enzo Life Sciences, Inc., USA

SO Eur. Pat. Appl., 4 pp.

CODEN: EPXXDW

DTPatent

LA English

FAN.CNT 1

	PATENT NO.	KIND DATE	APPLICATION NO.	DATE					
ΡI	EP 1348713	A2 20031001	EP 2003-4894	20030306					
	R: AT, BE, CH,	DE, DK, ES, FR,	GB, GR, IT, LI, LU, NL,	SE, MC, PT,					
	IE, SI, LT,	LV, FI, RO, MK,	CY, AL, TR, BG, CZ, EE,	HU, SK					
	US 2003225247	A1 20031204	US 2002-96075	20020312					
	CA 2421552	AA 20030912	20030912 CA 2003-2421552						
	JP 2004004048	A2 20040108	JP 2003-114988	20030311					
	US 2004203038	A1 20041014	US 2004-761906	20040121					
	US 2004254355	A1 20041216	US 2004-763076	20040122					
	US 2006172308	A1 20060803	US 2004-763088	20040122					
	US 2004176586	A1 20040909	US 2004-764418	20040123					
	US 2004192893	A1 20040930	US 2004-764417	20040123					
	US 2004230036	A1 20041118	US 2004-764389	20040123					
	US 2004229248	A1 20041118	US 2004-764393	20040123					

US 6949659 B2 20050927

US 2005004350 A1 20050106 US 2004-764388 20040123

PRAI US 2002-96075 A 20020312

AB This invention provides for labeling reagents, labeled targets and processes for preparing labeling reagents. The labeling reagents can take the form of cyanine dyes, xanthene dyes, porphyrin dyes, coumarin dyes or composite dyes. These labeling reagents are useful for labeling probes or targets, including nucleic acids and proteins. These reagents can be usefully applied to protein and nucleic acid probe based assays. They are also applicable to real-time detection processes.

IT 67987-16-0P 599177-71-6P

RL: ARU (Analytical role, unclassified); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation)

(labeling reagents and labeled targets, target

labeling processes and other processes for using same in

nucleic acid detns. and analyses)

RN 67987-16-0 CAPLUS

CN Phenanthridinium, 6,6'-[1,5-pentanediylbis(iminocarbonyl-3,1-phenylene)]bis[3,8-diamino-5-methyl-, dichloride (9CI) (CA INDEX NAME)

PAGE 1-A

●2 C1-

PAGE 1-B

-NH<sub>2</sub>

RN 599177-71-6 CAPLUS

CN Phenanthridinium, 6,6'-[1,5-pentanediylbis(iminocarbonyl-3,1-phenylene)]bis[3,8-diamino-5-methyl- (9CI) (CA INDEX NAME)

PAGE 1-A

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L9 ANSWER 4 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN
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AN 2003:734657 CAPLUS

DN 139:241306

 ${\tt TI}$  Methods and kits for detection of nucleic acids using fluorescence resonance energy transfer

IN Rabbani, Elazar; Stavrianopoulos, Jannis G.; Donegan, James J.; Coleman, Jack; Liu, Dakai

PA Enzo Life Sciences, Inc., USA

SO Eur. Pat. Appl., 115 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

ΙT

Pi	ATENT	NO.			KIND DATE					APP	LIC	DATE							
	EP 1344835 EP 1344835				A2 20030917 A3 20040331				EP	200		20030306							
	R: AT, BE, CH, IE, SI, LT,				DE, LV,	DK, FI,	•	•	•		•	•	•	LU, CZ,	•	SE, HU,	MC, SK	PT,	
U	S 2005	A1	. 2	2005	0623	US 2002-96076							20020312						
Ci	CA 2421556				AA	2	2003	0912	1	CA	200	3-2	2421	556		2	0030	311	
J1	P 2003	33340	97		A2	2	20031125 JP 2003-114989							20030311					
U	S 2006	50299	68		A1	2	20060209 US 2005-235516							20050926					
U	S 2006	50247	35		A1	2	2006	0202		US	200	5-2	2361	51		2	0050	927	
U	S 2006	50247	37		A1	2	2006	0202		US 2005-237442							20050927		
U	US 2006024738				A1	2	2006	0202		US	200	5-2	2374	67		2	0050	927	
US	US 2006035264				A1	A1 20060216			US 2005-237466							20050927			
PRAI U	AI US 2002-96076				Α	2	2002	0312											

AB This invention provides for compns. for use in real time nucleic acid detection processes. Such real time nucleic acid detection processes are carried out with energy transfer elements attached to nucleic acid primers, nucleotides, nucleic acid probes or nucleic acid binding agents. Real time nucleic acid detection allows for the qual. or quant. detection or determination of single-stranded or double-stranded nucleic acids of interest

in a sample. Other processes are provided by this invention including processes for removing a portion of a homopolymeric sequence, e.g., poly A sequence or tail, from an analyte or library of analytes. Compns. useful in carrying out such removal processes are also described and provided. 599177-71-6P

RL: ARG (Analytical reagent use); BUU (Biological use, unclassified); RCT (Reactant); SPN (Synthetic preparation); ANST (Analytical study); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(methods and kits for detection of nucleic acids using fluorescence resonance energy transfer)

RN 599177-71-6 CAPLUS

CN Phenanthridinium, 6,6'-[1,5-pentanediylbis(iminocarbonyl-3,1-phenylene)]bis[3,8-diamino-5-methyl- (9CI) (CA INDEX NAME)

PAGE 1-B

-NH<sub>2</sub>

L9 ANSWER 5 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 2001:671726 CAPLUS

DN 136:227333

TI Footprinting methods for analysis of pyrrole-imidazole polyamide/DNA complexes

AU Trauger, John W.; Dervan, Peter B.

CS Department of Chemistry, California Institute of Technology, Pasadena, CA, 91125, USA

SO Methods in Enzymology (2001), 340(Drug-Nucleic Acid Interactions), 450-466 CODEN: MENZAU; ISSN: 0076-6879

PB Academic Press

DT Journal; General Review

LA English

AB A review describes the footprinting methods for the anal. of pyrrole-imidazole polyamide/DNA complexes. Topics discussed include the preparation of 32P end-labeled DNA; preparation of polyamide serial dilns.; quant. DNase I footprinting procedure; methidiumpropyl-EDTA-Fe(II) footprinting protocol and affinity cleavage protocols; and characterization of eight-ring polyamide ImPy- $\beta$ -ImPy- $\gamma$ -ImPy- $\beta$ -ImPy- $\beta$ -Dp. (c) 2001 Academic Press.

IT 83789-87-1

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (footprinting methods for anal. of pyrrole-imidazole polyamide/DNA complexes)

RN 83789-87-1 CAPLUS

CN Iron, [3,8-diamino-6-[4-[13-(carboxy- $\kappa$ 0)-9,12-bis[(carboxy- $\kappa$ 0)methyl]-1,7-dioxo-2,6,9,12-tetraazatridec-1-yl- $\kappa$ N9, $\kappa$ N12]phenyl]-5-methylphenanthridiniumato(3-)]- (9CI) (CA INDEX NAME)

## RE.CNT 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 ANSWER 6 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1996:34743 CAPLUS

DN 124:66638

TI Composition for delivery of toxic radioisotopes to the cell nucleus for tumor therapy

IN Mattes, M. Jules

PA Center for Molecular Medicine and Immunology, USA

SO PCT Int. Appl., 23 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

	PATENT NO.						KIND DATE			APPLICATION NO.					DATE					
ΡI	WO	0 9529707				A1	_	19951109			WO 1995-US4440					19950421				
		W:	AM,	ΑT,	ΑU,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CZ,	DE,	DK,	EE,	ES,	FI,		
			GB,	GE,	HU,	IS,	JP,	ΚE,	KG,	ΚP,	KR,	ΚZ,	LK,	LR,	LT,	LU,	LV,	MD,		
			MG,	MN,	MW,	MX,	NO,	ΝZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	ТJ,		
		TT, UA																		
		RW:	ΚE,	MW,	SD,	SZ,	UG,	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IE,	ΙT,		
			LU,	MC,	ΝL,	PT,	SE,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	ML,	MR,	NE,		
			SN,	TD,	TG															
	CA	2189	051			AA 19951109					CA 1995-2189051					19950421				
	ΑU	9522	845			A1 19951129				AU 1995-22845						19950421				
	EΡ	7575	59			A1		1997	0212	EP 1995-916300						19950421				
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙE,	IT,	LI,	LU,	MC,	NL,	PT,	SE	
	JP	1050				Т2		1998							19950421					
	US	S 5759514				А	A 19980602			US 1996-695182										
PRAI	US	US 1994-235319				Α		1994	0429											
	WO 1995-US4440					W		1995	0421											

AB A conjugate of a tumor cell-targeting protein or polypeptide and a nucleic acid-targeting small mol. labeled with an Auger electron-emitting radionuclide is useful for tumor therapy. The tumor cell-targeting protein or polypeptide may be an antibody or fragment thereof, a hormone, or a growth factor. Thus, tumor-associated antigen-specific monoclonal antibody MA103 conjugated to 125I-labeled dilactitol-tyramine or 125I-labeled DTAF was internalized and processed by SK-RC-18 carcinoma cells, with retention half-lives of 104 and 52 h, resp. After lysosomal degradation of the antibody, the intracellularly released DTAF can pass through the lysosomal and nuclear membranes and bind to DNA.

IT 86388-76-3D, Methidium-spermine, radiolabeled, conjugates with tumor cell-binding proteins
RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(composition for delivery of toxic radioisotopes to cell nucleus for tumor therapy)

RN 86388-76-3 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[[[3-[[4-[(3-aminopropyl)amino]butyl]amino]propyl]amino]carbonyl]phenyl]-5-methyl-(9CI) (CA INDEX NAME)

$$_{\text{H}_{2}\text{N}}^{\text{NH}_{2}}$$
 $_{\text{Me}}^{\text{C-NH- (CH}_{2})} _{3} - _{\text{NH- (CH}_{2})} _{4} - _{\text{NH- (CH}_{2})} _{3} - _{\text{NH}_{2}}$ 

L9 ANSWER 7 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1993:59272 CAPLUS

DN 118:59272

TI A convenient method to synthesize tritium-labeled N-[3H]methyl-N-nitrosocarbamate transfer reagents

AU Mehta, Pratibha; Konakahara, Takeo; Gold, Barry

CS Eppley Inst. Res. Cancer Allied Dis., Univ. Nebraska, Omaha, NE, 68198-6805, USA

SO Journal of Labelled Compounds and Radiopharmaceuticals (1992), 31(11), 925-31 CODEN: JLCRD4; ISSN: 0362-4803

DT Journal

LA English

OS CASREACT 118:59272

AB Generally, the synthesis of activated nitrosocarbamates requires condensation of radiolabeled alkyl isocyanates with the appropriate alc. Radiolabeled alkyl isocyanates are com. unavailable and/or troublesome to synthesize; thus, an easy and economical method for preparing N-[3H]methyl-N-nitrosocarbamates suitable for use as transfer reagents via amidation of 1,2,2,2-tetrachloroethyl chloroformate with [3H]methylamine hydrochloride [affording 1,2,2,2-tetrachloroethyl N-[3H]methylcarbamate in 96% yield, sp. activity 30.1  $\mu$ Ci/mmol] has been developed.

IT 105885-55-0

RL: RCT (Reactant); RACT (Reactant or reagent)
 (carbamate-forming reaction of, with tritium-labeled
 tetrachloroethyl methylnitrosocarbamate)

RN 105885-55-0 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[{(2-aminoethyl)amino]carbonyl]phenyl]-5-methyl-, chloride (9CI) (CA INDEX NAME)

$$_{\text{H}_2\text{N}}^{\text{NH}_2}$$
 $_{\text{Me}}^{\text{C-NH-CH}_2-\text{CH}_2-\text{NH}_2}$ 

● cl-

ΙT 116405-70-0P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN116405-70-0 CAPLUS

Phenanthridinium, 3,8-diamino-5-methyl-6-[4-[[[2-CN

[[(methylnitrosoamino)carbonyl]amino]ethyl]amino]carbonyl]phenyl]-,

chloride (9CI) (CA INDEX NAME)

D Cl-

L9 ANSWER 8 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

ΑN 1990:18508 CAPLUS

DN 112:18508

TΙ Quantitative footprinting analysis of drug-DNA interactions: Fe(III) methidium-propyl-EDTA as a probe

ΑU Dabrowiak, James C.; Kissinger, Koren; Goodisman, Jerry

CS Dep. Chem., Syracuse Univ., Syracuse, NY, 13244-1200, USA

Electrophoresis (1989), 10(5-6), 404-12SO

CODEN: ELCTDN; ISSN: 0173-0835

DT Journal

LA English

AB Quant. footprinting studies involving a 139-base pair restriction fragment from pBR322 DNA, lexitropsin ligand and two different DNA cleavage agents, the enzyme DNase I and the footprinting reagent Fe(III) methidium-propyl-EDTA (Fe-MPE), are described. The autoradiog. data showed that the ligand, an analog of netropsin possessing two N-methylimidazole groups, binds to four regions on the 139-mer which are rich in GC. Anal. of the data leading to individual binding consts. for each of the four loading events on the 139-mer revealed that Fe-MPE and DNase I report the same binding consts. for the lexitropsin bound to its

interaction sequences. The fact that the data from both probes can be analyzed using a common model indicates that the DNA cleavage specificity of the probe and not its binding/cleavage mechanism is the important factor in reporting of site loading information in the footprinting experiment The study also showed that under certain conditions it is possible to gain information on the d. of ligand binding sites on carrier DNA by monitoring site loading events on the labeled fragment.

IT 122721-72-6 RL: ANST (Analytical study)

(in drug-DNA interactions study by quant. footprinting anal.)

RN 122721-72-6 CAPLUS

CN Iron(1+), [3,8-diamino-6-[4-[13-carboxy-9,12-bis(carboxymethyl)-1,7-dioxo-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methylphenanthridiniumato(3-)]-(9CI) (CA INDEX NAME)

L9 ANSWER 9 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1989:187936 CAPLUS

DN 110:187936

 ${\tt TI}$  Site-specific interaction of intercalating drugs with a branched DNA molecule

AU Guo, Qiu; Seeman, Nadrian C.; Kallenbach, Neville R.

CS Dep. Chem., New York Univ., New York, NY, 10003, USA

SO Biochemistry (1989), 28(6), 2355-9

CODEN: BICHAW; ISSN: 0006-2960

DT Journal

LA English

AΒ

The interaction of a stable branched DNA mol. with an intercalative drug is probed by hydroxyl radical scission. Methidiumpropyl-EDTA·Fe(II) [MPE·Fe(II)], consisting of an intercalating ring system tethered to EDTA·Fe(II), produces the hydroxyl radicals by a Fenton reaction. The cleavage patterns of each labeled strand in a branched tetramer of 4 16-mers are compared with those of the same strands in unbranched duplex controls. Strong differences between the profiles corresponding to scission of branched and duplex DNA mols. are seen in each of the strands at low MPE/DNA ratios. A specific site in the branched structure interacts preferentially with the drug, while other regions of the mol. are protected from cleavage. At  $4^{\circ}$ , cutting at strand positions demarcating the site of enhanced affinity is observed to be .apprx.60% more efficient than at the corresponding sequence positions in the control duplex DNA mols.; the degree of protection at the second site is comparable. Cleavage in the vicinity of the preferred site occurs at residues flanking the branch point that are asym. distributed with respect to it. The reactive Fe(II) group appears to be centered within 2 residues of the branch point, and the site of preferential intercalation may be between the 2 base pairs closest to the branch point in 1 of the 4 arms. The pattern of preferential cutting at this site is eliminated in the presence of excess propidium diiodide, another intercalative drug.

IT 83789-87-1

RL: BIOL (Biological study)

(DNA tetramer branched junction-containing and duplex forms intercalation of and cleavage by)

RN 83789-87-1 CAPLUS

CN Iron, [3,8-diamino-6-[4-[13-(carboxy- $\kappa$ 0)-9,12-bis[(carboxy- $\kappa$ 0)methyl]-1,7-dioxo-2,6,9,12-tetraazatridec-1-yl- $\kappa$ N9, $\kappa$ N12]phenyl]-5-methylphenanthridiniumato(3-)]- (9CI) (CA INDEX NAME)

L9 ANSWER 10 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1988:606515 CAPLUS

DN 109:206515

TI Synthesis of an N-methyl-N-nitrosourea linked to a methidium chloride analog and its reactions with phosphorus-32-end-labeled DNA

AU Konakahara, Takeo; Wurdeman, Richard L.; Gold, Barry

CS Eppley Inst. Res. Cancer Allied Dis., Univ. Nebraska, Omaha, NE, 68105, USA

SO Biochemistry (1988), 27(23), 8606-13 CODEN: BICHAW; ISSN: 0006-2960

DT Journal

LA English

OS CASREACT 109:206515

AΒ The synthesis and characterization of an N-methyl-N-nitrosourea (MNU) analog that is covalently linked to a methidium nucleus (I) is described. At 37° in pH 8.0 buffer I hydrolyzes via pseudo-first-order kinetics, with a calculated t1/2 = 77 min. By use of polyacrylamide sequencing gels the formation of piperidine-labile N7-methylguanine adducts from the reaction of I and MNU with 5'-32P-end-labeled DNA restriction fragments is reported. DNA methylation by I in 10 mM Tris buffer is enhanced with increasing ionic strength (50-200 mM NaCl), which contrasts to the inhibition of MNU-induced cleavage with increasing salt. In addition, I methylates all G sites equally, whereas MNU shows a clear preference for d(G)n ( $n \ge 3$ ) runs and an asym. methylation pattern within these G-rich regions. The results are discussed in terms of the delivery of the MNU moiety to the DNA target by a nonsequence-specific intercalation process and the subsequent hydrolytic generation of a nondiffusible alkylating intermediate.

IT 116375-33-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and hydrolysis of)

RN 116375-33-8 CAPLUS

$$H_2N$$
 $NH_2$ 
 $NH_2$ 
 $C-NH_2$ 
 $O$ 

● Cl-

IT 116405-70-0P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of and DNA methylation by)

RN 116405-70-0 CAPLUS

CN Phenanthridinium, 3,8-diamino-5-methyl-6-[4-[[[2-

[[(methylnitrosoamino)carbonyl]amino]ethyl]amino]carbonyl]phenyl]-, chloride (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} & \text{NH2} \\ & \text{H}_2\text{N} \\ & \text{Me} \\ & \text{C-NH-CH}_2\text{-CH}_2\text{-NH-C-N-Me} \\ & \text{O} \\ & \text{O} \\ \end{array}$$

● Cl -

IT 52671-19-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of and conversion to [[(aminoethyl)carbamoyl]phenyl]
derivative)

RN 52671-19-9 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-(4-carboxyphenyl)-5-methyl-, chloride (9CI) (CA INDEX NAME)

• c1-

IT 116375-34-9P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of and conversion to [[[[(nitrosomethylcarbamoyl)amino]ethyl]carbamoyl]phenyl] derivative)

RN 116375-34-9 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[[(2-aminoethyl)amino]carbonyl]phenyl]-5-methyl-, chloride, monohydrochloride (9CI) (CA INDEX NAME)

$$_{\text{H}_2\text{N}}^{\text{NH}_2}$$
 $_{\text{Me}}^{\text{C-NH-CH}_2-\text{CH}_2-\text{NH}_2}$ 

● c1-

● HCl

L9 ANSWER 11 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1988:161013 CAPLUS

DN 108:161013

TI Selective strand scission by intercalating drugs at DNA bulges

AU Williams, Loren Dean; Goldberg, Irving H.

CS Dep. Biol. Chem. Mol. Pharmacol., Harvard Med. Sch., Boston, MA, 02115, USA

SO Biochemistry (1988), 27(8), 3004-11 CODEN: BICHAW; ISSN: 0006-2960

DT Journal

LA English

AB Bulge-specific strand scission by the DNA intercalating/cleaving drugs neocarzinostatin chromophore (NCS-C), bleomycin (BLM), and methidiumpropyl-EDTA (MPE) is described. A series of 5'-32P end-labeled oligonucleotide duplexes were constituted that are identical except for the location of a bulge. In each successive duplex of the series, a bulge has been shifted stepwise up (from 3' to 5') one

strand of the duplex. Similarly, in each successive duplex of the series, sites of bulge-specific scission and protection were observed to shift in a stepwise manner. The results show that throughout the series of bulged duplexes, NCS-C causes specific scission at a site near a bulge, BLM causes specific scission at a site near a bulge, and MPE·Fe(II) causes specific scission centered around the bulge. Kn some sequences, NCS-C and BLM each cause bulge-specific scission at second sites. Further, bulged DNA shows sites of protection from NCS-C and BLM scission. The results are discussed with respect to a model of bulged DNA. It appears that specific scission at DNA bulges can be employed as a general assay for intercalation and binding orientation.

IT 90912-87-1

RL: BIOL (Biological study)

(DNA strand scission by, bulge-specific)

RN 90912-87-1 CAPLUS

CN Phenanthridinium, 3,8-diamino-6-[4-[13-carboxy-9,12-bis(carboxymethyl)-1,7-dioxo-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methyl-, chloride (9CI) (CA INDEX NAME)

PAGE 1-A

$$\begin{array}{c} \text{NH2} \\ \text{H2N} \\ \text{Me} \\ \text{O} \\ \text{O} \\ \text{O} \\ \text{O} \\ \text{O} \\ \text{O} \\ \text{CH2} - \text{CO}_2\text{H} \\ \text{CH2} - \text{CH2} - \text{N-CH2} - \text{CH2} \\ \text{O} \\$$

● Cl - .

PAGE 1-B

L9 ANSWER 12 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1983:466071 CAPLUS

DN 99:66071

TI Footprinting with MPE·iron(II). Complementary-strand analyses of distamycin- and actinomycin-binding sites on heterogeneous DNA

AU Van Dyke, M. W.; Dervan, Peter B.

CS Div. Chem. Chem. Eng., California Inst. Technol., Pasadena, CA, 91125, USA

SO Cold Spring Harbor Symposia on Quantitative Biology (1983), Volume Date 1982, 47(1), 347-53 CODEN: CSHSAZ; ISSN: 0091-7451

DT Journal

LA English

AΒ Complementary strand anal. of distamycin A (I)- and actinomycin D (II)-binding sites of Escherichia coli lactose operon DNA using the methidiumpropyl-EDTA-Fe(II) (III) footprinting technique was reported. Two restriction fragments, containing regions of identical sequence and 117 and 168 base pairs (bp) in length, were prepared with 3'-end 32P labeling on complementary strands and used as substrates. III cleavage inhibition patterns resulting from drug protection provided opposite-site footprints. Binding sites were localized on 50-bp sections of the restriction fragments. I binding protected 24 out of 50 bp, of which 20 (83%) were adenine-thymine pairs. The min. protected region was a 4-bp sequence, AATT. For this size sequence the binding d. (ligand bound/bp) was 0.12 or 6 mols. on the 50-bp sequence. II binding protected 22 out of 50 bp, 15 (68%) being guanine-cytosine pairs. The min. protected site was a 3-bp sequence, GTG, and the binding d. was 0.11-0.15 or .apprx.6-7 mols. on the 50-bp sequence. As predicted, the footprints were shifted 1-2 bp to the 3' side on each strand and were underprotected by 1 bp on the 5' side.

IT 83693-09-8

L9

RL: BIOL (Biological study)

(in DNA drug-binding complementary-strand anal., by footprinting)

RN 83693-09-8 CAPLUS

CN Iron, [3,8-diamino-6-[4-[13-(carboxy- $\kappa$ O)-9,12-bis[(carboxy- $\kappa$ O)methyl]-1-oxo-7-(oxó- $\kappa$ O)-2,6,9,12-tetraazatridec-1-yl]phenyl]-5-methylphenanthridiniumato(2-)]- (9CI) (CA INDEX NAME)

ΑN 1965:39544 CAPLUS

62:39544 DN

OREF 62:7002d-f

Preparation and biological activity of some complexes of trypanocidal TIphenanthridinium compounds

ΑU Groves, M. J.; Wilmshurst, E. C.

CS Boots Pure Drug Co. Ltd., Nottingham, UK

SO Journal of Pharmacy and Pharmacology (1964), 16(Suppl.), 140-6 CODEN: JPPMAB; ISSN: 0022-3573

DT Journal

LA English

AB Acacia, agar, carbopol 934, carboxymethyl cellulose, degraded carrageen, heparin, laminarin sulfate (I), pectin, Na alginate, stearic acid, sterculia, and tragacanth were complexed with Prothidium bromide (II), 2-amino-7-(2-amino-6-methyl-4-pyrimidylamino)-9-(pnitrophenyl)phenanthridinium 10,1'-dimethochloride (III), 2-amino-7-(2-amino-6-methyl-4-pyrimidylamino)-9-phenylphenanthridinium 10-ethomethanesulfonate 1'-methomethanesulfonate (IV), homidium bromide, isometamidium, and dimidium. The insol. complexes were prepared by slow addition of an excess of 1% solution of phenanthridinium salt in water, with stirring, to a 1% solution or gel of the complexing substance. The mixture was stirred, let stand 1 hr., then centrifuged. The precipitate was resuspended

and

recentrifuged. After total N analysis the precipitate was suspended in 2% hydroxyethyl cellulose. Soluble solns. were also prepared Saturated solns. of II

or homidium bromide added to agar or pectin produced a color change but no precipitate The agar complexes were evaporated to dryness then redissolved in

hot

Pectin complexes were redissolved in excess of EtOH. Mice were injected subcutaneously with 0.2 or 1.0 mg. of phenanthridinium compound/kg. or the complexes. The mice were then challenged with injections of Trypanosoma congolense. Only complexes of II with I showed any marked activity in the mouse. The resuspended II-I complex was less active than the original suspension. I and degraded carrageen enhanced the activities of isometamidium and III. Coupling with I enhanced IV activity. Complexing did not change the effect of II against trypanosomiasis in cattle.

20566-69-2, Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl-ΙT (salts, reaction products with polysaccharides, preparation of and Trypanosoma response to)

20566-69-2 CAPLUS RN

Phenanthridinium, 3,8-diamino-5-methyl-6-phenyl- (8CI, 9CI) CNNAME)